

Structural and luminescent properties of europium doped TiO₂ thick films synthesized by the ultrasonic spray pyrolysis technique

E Zaleta-Alejandre¹, M Zapata-Torres¹, M García-Hipólito²,
M Aguilar-Frutos¹, G Alarcón-Flores¹, J Guzmán-Mendoza¹ and
C Falcony³

¹ Centro de Investigación en Ciencia Aplicada y Tecnología Avanzada-IPN; Legaria # 694, Col. Irrigación, Del. Miguel Hidalgo, México D.F.

² Instituto de Investigaciones en Materiales-Universidad Nacional Autónoma de México; A.P. 70-360, Coyoacán 04510, México D.F.

³ Centro de Investigación y de Estudios Avanzados-IPN; Departamento de Física, Apdo. Postal 14-470, Del. Gustavo A. Madero, C.P. 07000, México, D.F.

E-mail: ezaletaa@ipn.mx

Received 16 December 2008, in final form 14 February 2009

Published 15 April 2009

Online at stacks.iop.org/JPhysD/42/095102

Abstract

The structural and luminescent properties of trivalent europium-doped titanium dioxide films synthesized by the ultrasonic spray pyrolysis technique at several substrate temperatures are reported. These films are nanocrystalline and present a mixture of tetragonal (anatase and rutile) crystal structures of the titania as determined by x-ray diffraction. The rutile crystal structure became predominant as the substrate temperature during deposition was increased. Under UV and electron beam excitation, these coatings showed strong luminescence due to f-f transitions and the dominant transition was the hypersensitive $^5D_0 \rightarrow ^7F_2$ red emission of Eu³⁺. The photo- and cathodoluminescence characteristics of these films were studied as a function of growth parameters such as substrate temperature and europium concentration. Excitation with a wavelength of 396 nm resulted in photoluminescent emission peaks located at 557, 580, 592, 615, 652 and 703 nm, associated with the electronic transitions of the Eu³⁺ ion. The photoluminescence (PL) intensity as a whole is observed to decrease as the deposition temperature is increased. Also, with increasing doping concentration, a quenching of the PL is observed. The chemical composition and surface morphology characteristics of the films are also reported.

1. Introduction

Titanium dioxide (TiO₂) is one of the most extensively studied metal oxides because of its potential applications as photo-catalyzer, self-cleaning material, ultrawhite pigment, light scatterer, bio-compatible material and high refractive material for optoelectronic devices [1–5], and also in waste water purification [6], in solar cells [7], as a gas sensor material [8], host lattice for phosphors [9] and thermoluminescent applications [10], among others. TiO₂

films have been deposited by different methods, such as ablation laser deposition [11], RF-sputtering [12], sol-gel processes [13], chemical vapour deposition (CVD) [14] and ultrasonic spray pyrolysis (USP) [15]. The USP technique is considered as a low cost process, versatile, easily scalable to industrial applications and appropriate for the deposition of metallic oxide films doped with rare earth ions [16, 17].

TiO₂ has also attracted much attention for many years due to its physical properties, which show variations influenced by impurities, oxygen defects and crystalline modifications. In